

## Electrostatic Charging

### Equipment List

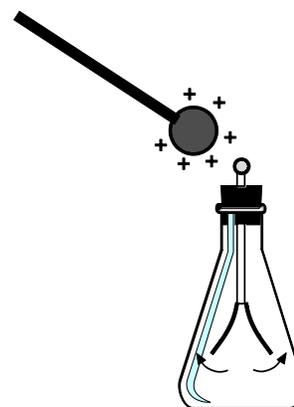
Qty	Items	Part Numbers
1	Charge Sensor	CI-6555
1	Charge Producers and Proof Planes	ES-9057A
1	Faraday Ice Pail	ES-9024A

### Introduction

The purpose of this activity is to investigate the nature of charging an object by contact as compared to charging an object by induction. You will also determine the polarity of two charge ‘producers’ and measure the amount of charge on each.

### Background

Electric charge is one of the fundamental properties of matter. Electrostatics is the study of electric charges and their characteristics. For example, like charges tend to repel and unlike charges attract. An object is electrically neutral most of the time; that is, it has a balance of positive and negative electric charges. The positive charges (+) come from the proton, while the negative charges (-) are a result from the electrons. Rubbing different materials together, contact with a charged object, and charging by induction are the three ways to create an imbalance of electric charge – sometimes called static electricity. Static electricity is a charge and the unit of charge is the coulomb with its SI symbol,  $q$ . Any positive or negative charge,  $q$ , that can be detected can be written as  $q = ne$  where  $n = \pm 1, \pm 2, \pm 3, \dots$  in which  $e$ , the **elementary charge**, has the value of  $1.602 \times 10^{-19}\text{C}$ . Frequently in experiments milliCoulombs (mC), microCoulombs ( $\mu\text{C}$ ), nanoCoulombs (nC) and even picoCoulombs (pC) are used.



As mentioned above, opposite charges always attract and like charges tend to repel. At an elemental level, like charges always repel (electrons repel electrons, protons repel protons), but for macroscopic objects, non-symmetric charge distribution can result in an overall attraction between two objects that carry the same type of overall charge (positive or negative). Non-symmetrical charge distribution always results in an attraction between a charged object and an electrically neutral (overall) object. Looking at the three types of charging mentioned earlier, we can look at how the charges are distributed in each case:

1. *Charging by rubbing*: when two initially neutral non-conducting objects are rubbed together, one of them will generally bind electrons more strongly than the other and take electrons from the other. The law of conservation of charge requires that the total amount of electrons be conserved. That is, electrons only move from one object to another, but no new electrons are created, nor do they disappear. Overall, the two objects when considered together still have zero net charge.
2. *Charging by contact*: when a charged object is touched to a neutral (or less charged) object, repulsive forces between the like charges result in some of the charge transferring to the less

charged object so the like charges will be further apart. This effect is much larger for conducting objects.

3. *Charging by induction*: the protons and electrons inside any object respond to electric forces of attraction or repulsion. When an object is placed near a charged object, the charged object will exert opposite forces on the protons and the electrons inside the other object, forcing them to move apart from each other. One side of the object will become more positive than it was initially. The other side will become more negative, as electrons migrate internally. This condition is called **polarization**, a word that refers to the object having “poles,” or opposite sides with different electrical states, even though the object as a whole may still neutral. If a conductor is touched to the polarized object, some of the charge will transfer to the conductor. If the conductor is then removed, the object now carries a net charge different from its initial charge.

Historically, Michael Faraday used a metal ice pail as a conducting object to study how charges distributed themselves with a charged object was brought inside the pail. The ‘ice pail’ had a lid with a small opening through which he lowered a positively-charged metal ball into the pail without touching it to the pail. Negative charges in the pail moved to the inner surface of the pail leaving positive charges on the outside.

If the charged ball touches the inside of the ice pail, electrons would flow into the ball exactly neutralizing the ball. This would leave the pail with a net positive charge residing on the outer surface of the pail.

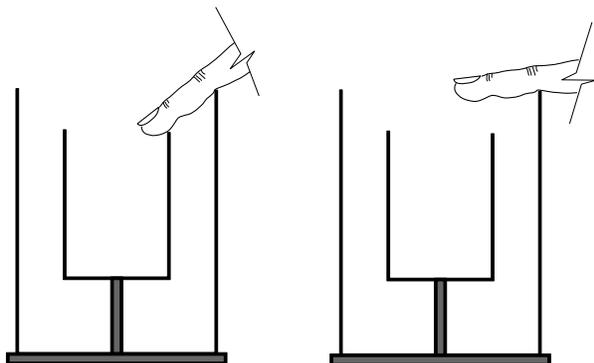
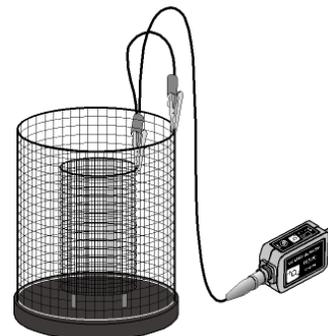
To experimentally investigate electrostatics, some charge-detecting or measuring device is needed. A common instrument for this purpose is the electroscope, a device with two thin gold leaves vertically suspended from a common point. When a charged object is brought near the electroscope, the gold leaves separate, roughly indicating the magnitude of the charge.

Although there are many different versions of the electroscope, all such instruments depend upon the repulsion of like charges to produce an output or reading. Unfortunately, such devices are relatively insensitive (large amounts of charge are needed to make the gold leaves separate), and the device does not have a quantitative reading.

The Charge Sensor is an ‘electronic electroscope’. In addition to providing a quantitative measurement, the Charge Sensor is more sensitive and indicates polarity directly. Assuming there is no residual charges and no charge leakage in the experiment, this instrument should provide accurate results.

## Setup

1. Connect the Charge Sensor to the interface & start *PASCO Capstone*.
2. Set the “Gain Select Switch” of the Charge Sensor to 5X and the sample rate to 10 Hz.
3. Open a graph displaying the voltage from the Charge Sensor (Charge vs. Time Graph) and a Meter display.
4. Connect the alligator clips of the sensor’s cable assembly to the inner (**longer wire inside – red band**) and outer baskets (**shorter wire outside – black band**) of the Faraday Ice Pail.



### Preparing to Record Data

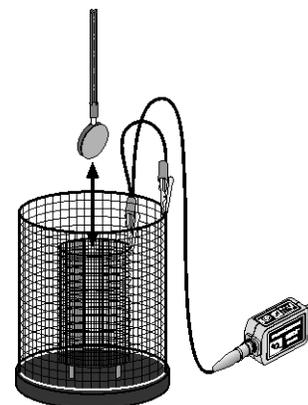
Before starting any experiment using the ‘Faraday Ice Pail’, the pail must be momentarily grounded. To ground the pail, touch the inner pail and the shield at the same time with the finger of one hand as shown in the image to the left.

**NOTE:** You may need to redo the grounding and/or zeroing of the Ice Pail during the experiment. It is very easy to transfer charge to the ice pail by touching it or even getting too close to it with a charged object. It may even acquire a charge sitting on the table for a while. To see how sensitive the system is, stick a finger down the axis of the inner cylinder (without touching the cylinder.) Now rub your fingers through your hair, or on your shirt, or shuffle your shoes on the floor and try sticking your finger back into the Ice Pail. See any difference? What happens if you touch the Ice Pail? What’s the moral about where you put your hands during the experiment? **Redo the grounding of the Ice Pail immediately before each procedure.**

## Procedure

### Part 1: Determine the Polarity of the Charge Producers

1. Ground the ‘Ice Pail’ and press the ‘ZERO’ button on the Charge Sensor to discharge the sensor. Insert, without touching the walls, the charge producers independently and together into the inner pail. Record their values for Question 1.
2. Briskly rub the dark charge producer and white Charge Producers together several times, holding them as close to the bottom as possible.
3. Click ‘Record’ in *Capstone* to start recording data.
  - Without touching the ‘Ice Pail’, lower the white Charge Producer into the ‘Ice Pail’. Watch the Meter and Graph displays.
  - Remove the white Charge Producer and then lower the Dark Charge Producer into the ‘Ice Pail’. Watch the results.



4. After a few moments remove the Dark Charge Producer and stop recording data.

**Part 2: Measure the Charge on the White Charge Producer.**

5. Ground the 'Ice Pail' and press the 'ZERO' button on the Charge Sensor to discharge the sensor.
6. Briskly rub the Dark and white surfaces of the Charge Producers together several times.
7. Start recording data. Record the zero value on line 1.
  - Lower the white Charge Producer into the 'Ice Pail'. Record this value on line 2.
  - Rub the surface of the white Charge Producer against the inner pail Watch the Meter and Graph displays. Record this value on line 3.
  - Remove the Charge Producer from the inner pail. Record this value on line 4.
8. After a few moments, stop recording data and save graph.

**Measure the Charge on the Dark Charge Producer**

9. Ground the 'Ice Pail' and press the 'ZERO' button on the Charge Sensor to discharge the sensor.
10. Briskly rub the Dark and white surfaces of the Charge Producers together several times.
11. Start recording data. Record the zero value on line 1.
  - Lower the Dark Charge Producer into the 'Ice Pail'. Record this value on line 2.
  - Rub the surface of the Dark Charge Producer against the inner pail Watch the Meter and Graph displays. Record this value on line 3.
  - Remove the Charge Producer from the inner pail. Record this value on line 4.
12. After a few moments, stop recording data and save graph.

**Part 3: Charge the 'Ice Pail' by Induction**

13. Ground the 'Ice Pail' and press the 'ZERO' button on the Charge Sensor to discharge the sensor.
14. Briskly rub the Dark and white surfaces of the Charge Producers together several times.
15. Start recording data. Record the zero value on line 1.
  - Without touching the 'Ice Pail' with the Charge Producer, lower the white Charge Producer into the 'Ice Pail'. Record this value on line 2.
  - While the Charge Producer is still inside the inner pail, use the finger of one hand to momentarily ground the 'Ice Pail'. Watch the results, the charge should go to approximately zero. Record this value on line 3.
  - After you ground the 'Ice Pail', remove your hand and then remove the Charge Producer.
16. After a few moments, record this value on line 4 and stop recording data.
17. Ground the 'Ice Pail' and zero the sensor and repeat the procedure using the Dark Charge Producer and save graph.

**Lab Report: Electrostatic Charging**

Name: \_\_\_\_\_

**Graphs:**

Indicate on the graphs each step of the experiment, i.e. charge producers ‘far’ away from experiment; charge producer inducing current, pail grounded, charge producer in contact with pail, etc. You can annotate directly on the graph by using the large red A:  on the graph display, or physically write on the graphs once printed out.

Line	State	White Disk Contact ( $\mu\text{C}$ )	Dark Disk Contact ( $\mu\text{C}$ )	White Disk Induced ( $\mu\text{C}$ )	Dark Disk Induced ( $\mu\text{C}$ )
1	Zero				
2	Initial				
3	Transfer				
4	Disk Out				

**Analysis****Questions**

- From Part 1, what charge and polarity (what sign) are the two Charge Producers? Where did the electrons go (or come from)?
- (Part 1) After the rubbing process, what is the overall net charge of the two wands?
- (Part 1) What evidence is there that the rubbing gave each wand an electrical charge?

4. (Part 2) After charging by contact, what was the charge of the white charge producer and the basket? Was it the same polarity?
  
5. (Part 2) After charging by contact, what was the charge of the dark charge producer and the basket? Was it the same polarity?
  
6. (Part 3) After charging by induction, what was the charge of the white charge producer and the basket? Was it the same polarity?
  
7. (Part 3) After charging by induction, what was the charge of the dark charge producer and the basket? Was it the same polarity?
  
8. How does the process of charging by contact differ from the process of charging by induction?
  
9. How many electrons are inside of the pail if the pail has a charge of  $50 \mu\text{C}$ ?
  
10. Assuming you have  $6.24 \times 10^{14}$  electrons and the surface area of the pail is  $0.2 \text{ m}^2$ , what is the charge density ( $\text{C}/\text{m}^2$ )?

11. Previously experiments were conducted where two pith balls were used to look at static electricity. Looking at one pith ball as an example, shown below, show the mathematical proof that in this equation the tan can be replaced with a sin. (Hint: Small angle approximation)

$$F_c = mg \tan \beta \cong mg \sin \beta = \frac{mgx}{l}$$

